

CHEMISTRY AND CHEMICAL ENGINEERING

Volume 2024 | Number 3

Article 4

September 2024

STUDY OF CRYSTALLIZATION ABILITY OF GLASSES SYNTHESIZED IN THE BASALT-QUARTZ-SODA SYSTEM

Eldor I. RUZMATOV

Tashkent Institute of Chemical Technology, Tashkent, Uzbekistan, eldor_bbk@mail.ru

Mastura Kh. ARIPOVA

Tashkent Institute of Chemical Technology, Uzbekistan, aripova1957@yandex.com

Bahrom R. Ro'ziboyev

Tashkent Institute of Chemical Technology, Tashkent, Uzbekistan, bahrom_smd@mail.ru

Follow this and additional works at: <https://cce.researchcommons.org/journal>



Part of the Materials Science and Engineering Commons

Recommended Citation

RUZMATOV, Eldor I.; ARIPOVA, Mastura Kh.; and Ro'ziboyev, Bahrom R. (2024) "STUDY OF CRYSTALLIZATION ABILITY OF GLASSES SYNTHESIZED IN THE BASALT-QUARTZ-SODA SYSTEM," *CHEMISTRY AND CHEMICAL ENGINEERING*: Vol. 2024: No. 3, Article 4.

DOI: <https://doi.org/10.34920/cce202434>

Available at: <https://cce.researchcommons.org/journal/vol2024/iss3/4>

This Article is brought to you for free and open access by Chemistry and Chemical Engineering. It has been accepted for inclusion in CHEMISTRY AND CHEMICAL ENGINEERING by an authorized editor of Chemistry and Chemical Engineering. For more information, please contact zuchra_kadirova@yahoo.com.

STUDY OF CRYSTALLIZATION ABILITY OF GLASSES SYNTHESIZED IN THE BASALT-QUARTZ-SODA SYSTEM

Eldor I. RUZMATOV (eldor_bbk@mail.ru),
Mastura Kh. ARIPOVA (arianova1957@yandex.com)
Bakhram R. RUZIBAYEV (bahrom_smd@mail.ru)
Tashkent Chemical-Technological Institute, Tashkent, Uzbekistan

The aim of the work is to study the crystallization capacity of glasses synthesized in the basalt-quartz-soda system. The crystallization capacity of glasses was studied by the method of mass crystallization in the range of 600 °C and 1050 °C. The stages of spontaneous crystallization were determined by the DTA method. The density, TCLE, refractive index, compressive strength and frost resistance were studied. The crystalline phases of crystallized glasses were determined by the X-ray phase method. The presence of diopside and anorthite was revealed.

Keywords: colored glass, basalt, crystallization, temperature, properties

ИССЛЕДОВАНИЕ КРИСТАЛЛИЗАЦИОННОЙ СПОСОБНОСТИ СТЕКОЛ, СИНТЕЗИРОВАННЫХ В СИСТЕМЕ БАЗАЛЬТ-КВАРЦ-СОДА

Эльдор И. РУЗМАТОВ (eldor_bbk@mail.ru),
Мастура Х. АРИПОВА (arianova1957@yandex.com)
Бахрам Р. РУЗИБАЕВ (bahrom_smd@mail.ru)
Ташкентский химико-технологический институт, Ташкент, Узбекистан

Целью работы является исследование кристаллизационной способности стекол, синтезированных в системе базальт-кварц-сода. Исследована кристаллизационная способность стекол методом массовой кристаллизации в интервале 600 °C и 1050 °C. Определены этапы самопроизвольной кристаллизации методом ДТА. Изучены плотность, ТКЛР, показатель преломления, прочность на сжатие и морозостойкость. Рентгенофазовым методом определены кристаллические фазы закристаллизованных стекол. Выявлено наличие кристаллов диопсида и анортита.

Ключевые слова: цветные стекла, базальт, кристаллизация, температура, свойства

BAZALT-KVARTS-SODA TIZIMIDA SINTEZ QILINGAN SHISHALARING KRISTALLANISH QOBILIYATINI O'RGANISH

Eldor I. RUZMATOV (eldor_bbk@mail.ru),
Mastura X. ARIPOVA (arianova1957@yandex.ru)
Baxram R. RUZIBAYEV (bahrom_smd@mail.ru)
Toshkent kimyo-tehnologiya instituti, Toshkent, O'zbekiston

Ishning maqsadi bazalt-kvarts-soda tizimida sintez qilingan shishalarining kristallanish qobiliyatini o'rganishdir. 600 °C va 1050 °C oralig'ida massaviy kristallanish usuli bilan shishalarining kristallanish qobiliyati o'rGANildi. DTA tahlili yordamida shishalarining nukleatorlar sifatida kristallanish qobiliyati aniqlandi. Shishakristall namunalarning zichligi, TCHKK, nur sindirish ko'ssatkichi, mexanik siqilishga mustaxkamlik darajasi va sovuqqa bardoshliligi o'rganildi. Rentgen tahlili yordamida kristallangan shishalarining kristall fazalari o'rganilib, diopsid va anortitning kristallari mayjudligi aniqlandi.

Kalit so'zlar: rangli shishalar, bazalt, kristallanish, harorat, xossalalar

DOI: [10.34920/cce202434](https://doi.org/10.34920/cce202434)

Kirish

Shisha ishlab chiqarish texnologiyasi tabiiy va texnik xomashyolardan foydalangan holda turli sohalar uchun kerakli bo'lgan estetik jihatdan nafis va bezakbop materiallarni sintez qilish imkoniyatini ochadi. Shishaning kashf etilishi tufayli yillar davomida turli shakldagi idishlar, tarali shishalar, deraza oynalari qisqacha qilib aytganda, jamiyat turmush tarzi uchun zarur bo'lgan buyumlarni ko'plab ishlab chiqarilishiga olib keldi [1-5].

Hozirgi vaqtida dunyoda zamonaviy va modernizatsiya qilingan binolar hamda inshootlarni qurishda ichki va jabhani dekorativ jihatdan bezash muhim ahamiyatga ega. Arxitekturada binolarning tashqi va ichki dizaynnini estetik hamda funksional yaxshilashda shisha buyumlardan foydalanish asosiy omillardan biri hisoblanadi [6-8].

Shishalarни bir yoki ikki bosqichda termik ishlov berish orqali yuqori xususiyatarga ega

bo'lgan shishakristall materiallar olish mumkin [9-12]. Maxsus texnologiya yordamida yuqori mexanik mustahkamlikga, termik kengayish koeffisienti darajasi past va tashqi ta'sirlarga yuqori qarshilikga ega bo'lgan materiallar shishakristall materiallar deb ataladi [13-14]. Tog' jinslari asosida olingan shishakristall materiallar yuqori fizik-kimyoviy xossalarga ega bo'lib ulardan binolarning tashqi fasadlarini bezashda yoki zinapoyalarni qoplashda hamda pol uchun koshinlar sifatida foydalaniadi [15-17]. Yer yuzida eng ko'p tarqalgan va boshqa xomashyolarga nisbatan arzon bo'lgan basalt tog' jinsi shishakristall materiallar ishlab chiqarishda keng qo'llaniladi [18-20].

Ushbu tadqiqotning maqsadi basalt-kvarts-soda tizimida sintez qilingan optimal tarkibli shishalarining kristallanish qobiliyatlarini o'rganishdan iborat. Shishalar sintezi uchun Osmonsoy koni bazalti, Samarqand kvarts qumi va Qo'ng'iroq texnik sodasi qo'llanilgan.

1-jadval

Basalt-kvarts-soda tizimida sintez qilingan shishalarning kimyoviy tarkibi

Shisha indeksi	Oksidlarning miqdori, %												
	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	CaO	MgO	Na ₂ O	K ₂ O	TiO ₂	Cr ₂ O ₃	ZnO	MnO	CuO	SO ₃
2	55,6	11,7	7,78	7,48	3,06	10,2	0,95	1,83	0,015	0,0116	0,108	0,007	0,104
3	59,5	10,1	6,91	6,74	2,71	9,95	1,14	1,62	0,014	0,010	0,097	0,005	0,070
9	65,0	8,77	5,85	5,71	2,28	9,46	0,98	1,37	0,012	0,008	0,081	0,006	0,057
11	71,5	3,58	1,25	1,36	0,41	20,0	1,21	0,32	0,003	0,002	0,019	0,002	0,059
15	58,8	6,04	3,80	3,75	1,43	21,0	0,97	0,88	0,007	0,0055	0,054	0,004	0,098
20	49,8	12,8	6,84	7,62	3,10	17,0	0,60	1,71	0,010	0,008	0,099	0,006	0,126
21	50,9	13,2	8,53	9,61	3,33	10,8	0,69	2,14	0,011	0,012	0,116	0,006	0,151
22	52,5	10,5	7,11	6,23	2,81	17,6	0,81	1,74	0,013	0,008	0,097	0,006	0,074
23	55,7	8,13	5,86	5,98	2,20	15,4	0,87	1,36	0,012	0,009	0,054	0,005	0,181
26	44,0	10,8	3,71	3,72	1,50	24,5	0,88	0,88	0,007	0,005	0,055	0,004	1,380
32	72,4	4,29	2,37	2,48	0,85	15,0	0,98	0,57	0,005	0,005	0,036	0,003	0,057
33	72,8	4,37	2,44	2,47	0,86	15,1	0,98	0,56	0,004	0,006	0,038	0,003	0,050

Tadqiqot usullari

Shishakristall namunalarning fazaviy tahlili CuK yordamida XRD-6100 apparatida (Shimadzu, Yaponiya) qayd etilgan diffraktsion tasvirlar asosida amalga oshirildi. - radiatsiya (b - filtr, Ni, 1.54178 quvur oqimi va kuchlanish rejimi 30 mA, 30 kV) va doimiy detektor aylanish tezligi 4 daraja / min 0,02 daraja qadam bilan va skanerlash burchagi 4 dan 80 ° gacha o'zgarib turadi.

Shishakristall namunalarning fizik-kimyoviy xossalari GOST 13996-2019 talablari bo'yicha aniqlandi

Natijalar va muhokama

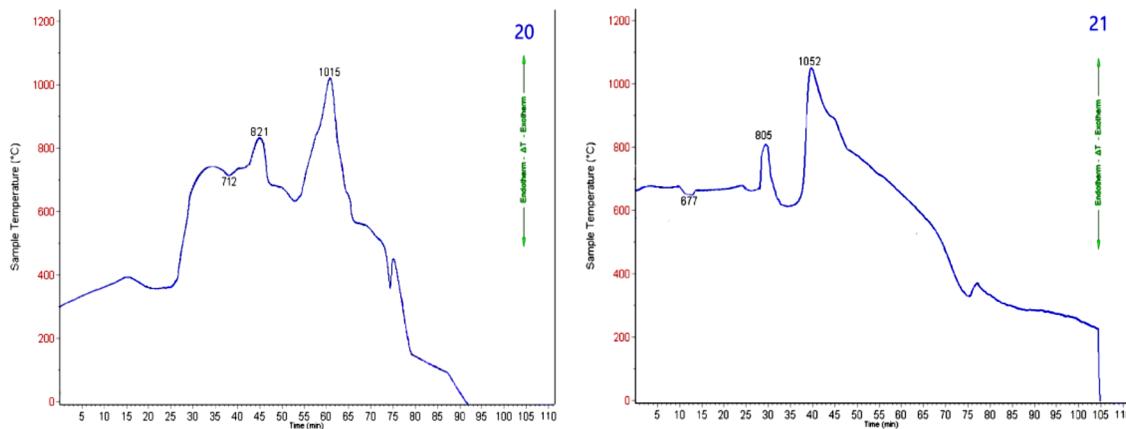
Basalt-kvarts-soda tizimida sintez qilingan shishalarning kimyoviy tarkibi 1-jadvalda keltirilgan.

1-jadvaldagи ma'lumotlardan ko'rilib turibdiki, basalt-kvarts-soda tizimida sintez qilingan shishalarning kimyoviy tarkibida Fe₂O₃, TiO₂, Cr₂O₃, va MnO rang beruvchilarning mavjudligi jigarang, qora, firuza, yashil, och havorang va sariq rangli shishalar olish imkonini bergen [21]. Olingan shishalarning fizik-kimyoviy xossalari quyidagi 2-jadvalda keltirilgan.

2-jadval

Bazalt-kvars-soda tizimida sintez qilingan optimal tarkibli shishalarning fizik-kimyoviy xossalari

Shisha indeksi	Zichligi, kg/m ³	TCHKK •10 ⁷ , K ⁻¹	Nur sindirish ko'rsatkichi, n	Mexanik egilishga mustahkamlik darajasi, MPa	Sovuqqa bardoshliligi, tsikl
2	2743	84	1,471	50	37
3	2694	83	1,470	52	37
9	2675	79	1,468	49	38
11	2481	112	1,466	35	34
15	2621	119	1,462	37	33
20	2760	112	1,476	38	35
21	2766	87	1,474	55	40
22	2728	110	1,474	34	34
23	2676	104	1,482	36	35
26	2648	94	1,477	44	36
32	2584	115	1,464	33	33
33	2512	88	1,476	41	36
GOST P 54169-2010	2480 – 2700	80 – 120	1,4 – 1,5	15 – 45	30-35



1-rasm. Bazalt-kvars-soda tizimida oling shisha namunalarining termogrammasi.

2-jadvaladagi ma'lumotlardan bazalt-kvars-soda tizimida sintez qilingan optimal tarkibli shishalarning fizik-kimyoviy xossalari GOST P 54169–2010 talablariga to'liq javob berishi aniqlandi.

Ma'lumki shishakristall materiallar olishda shishalarning kritsallanish bosqichlari muhim ahamiyatga ega bo'lib, bu jarayon differential termik tahlil usuli orqali aniqlanadi [22]. Shu maqsadda bazalt-kvars-soda tizimida sintez qilingan shishalarning kristallanish haroratlari aniqlan-

di va natijalar quyidagi 1-rasmda keltirilgan.

1-rasmdagi natijalarga ko'ra, shisha kritsallanishi ikki bosqichda sodir bo'ladi deb xulosa qilindi: 600–800 °C harorat oralig'ida birinchi bosqich kristall markazlarning shakllanishi va ikkinchi bosqich – taxminan 1050 °C haroratda kritsallanishning o'sishi. Olingan tahlil natijalarini hisobga olgan holda, shishalarning kristallanish qobiliyati massaviy kristallanish usuli yordamida o'rGANildi (3-jadval).

3-jadval

Bazalt–kvars–soda tizimida olingan shishalarning kristallanish qobiliyati

Shisha indeksi	Harorat ta'sirida kristallanish darajasi					
	600 °C	700 °C	800 °C	900 °C	1000 °C	1050 °C
2	[Image]	[Image]	[Image]	[Image]	[Image]	[Image]
3	[Image]	[Image]	[Image]	[Image]	[Image]	[Image]
9	[Image]	[Image]	[Image]	[Image]	[Image]	[Image]
11	[Image]	[Image]	[Image]			
15	[Image]	[Image]	[Image]			
20	[Image]	[Image]	[Image]	[Image]	[Image]	[Image]
21	[Image]	[Image]	[Image]	[Image]	[Image]	[Image]
22	[Image]	[Image]	[Image]	[Image]	[Image]	[Image]
23	[Image]	[Image]	[Image]	[Image]	[Image]	
26	[Image]	[Image]	[Image]	[Image]	[Image]	
32	[Image]	[Image]	[Image]	[Image]		
33	[Image]	[Image]	[Image]			

- Kristallanish turi:
- [Image] - kristallanish belgilari yo'q
 - [Image] - kichik ajratilgan sohalarda sirt tomondan kristallanish
 - [Image] - uzuksiz yupqa
 - [Image] - qoplam shaklida sirt kristallanishi
 - [Image] - qisman taqsimlangan qalin qobiq shaklida yuza kristallanishi
 - [Image] - kristallanishning namunada butun hajm bo'ylab tarqalishi -
 - [Image] - shishalarda erish jarayonining boshlanishi
 - [Image] - to'liq hajmli kristallanish.

Shishalarning kristallanish qobiliyati 600–1050 °C harorat oralig‘ida issiqlik bilan ishlov berish orqali o‘rganildi, ushlab turish vaqtiga 1 soatni tashkil etdi. Tajribalar natijasida 2, 3, 9, 20 va 21-tarkibli shisha namunalarini 1050 °C haroratda to‘liq kristallanishi aniqlandi. 22-tarkibli shisha namunasi 1000 °C haroratda shartli ravishda to‘liq kristallandi va harorat oshishi bilan erish jarayoni kuzatildi.

Bazalt-kvars-soda tizimida sintez qilingan shishalarning kristallanish qobiliyatini tahlil qilish natijasida 700 °C haroratda shisha namunalarida kristall markazlarning shakllanishi va 1050 °C haroratda to‘liq kristallanishi aniqlandi.

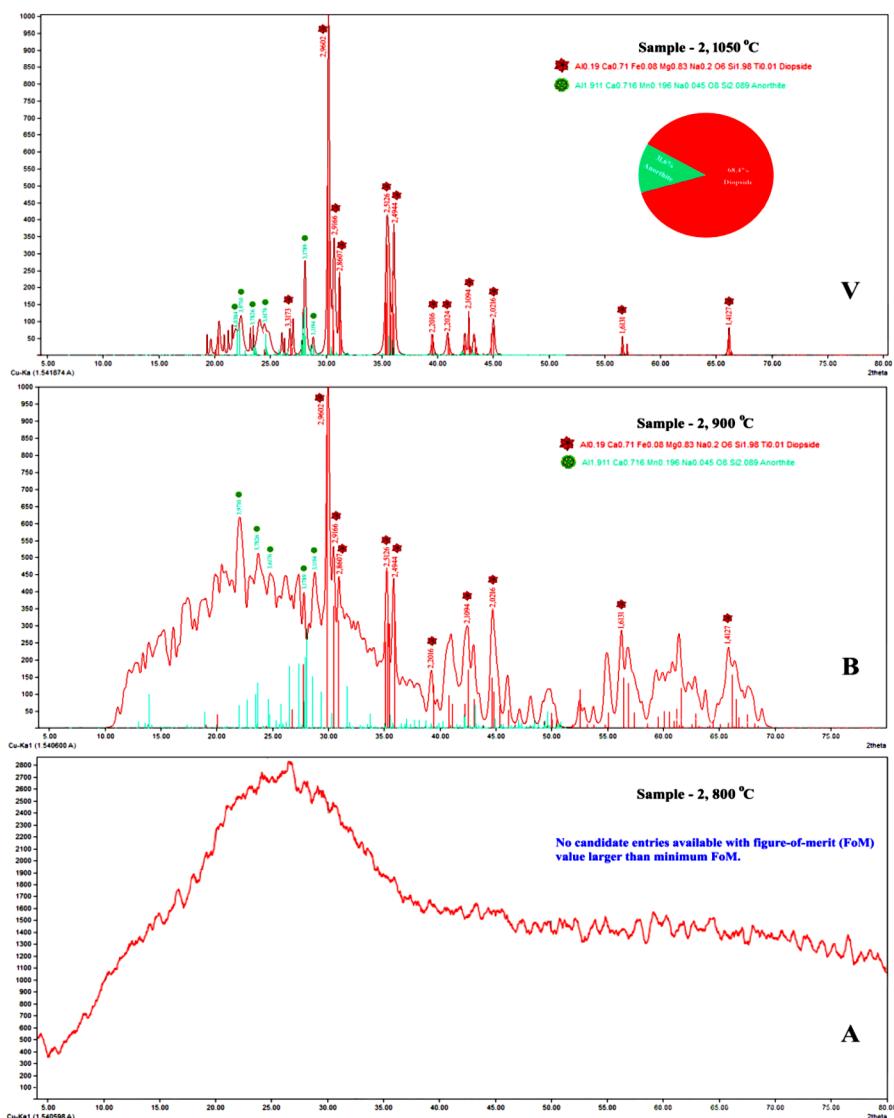
Rentgen tahlil natijalari shuni ko‘rsatdiki, 2-tarkibli shisha namunalariga 800 °C, 900 °C va 1050 °C haroratlarda issiqlik bilan ishlov berish jarayonida faza o‘zgarishlari sodir bo‘lgan - amorf fazaning to‘liq yo‘qolishi va harorat 1050 °C

yetganda diopsid va anortit fazalarining hosil bo‘lishi kuzatildi (2-rasm). Diopsid kristall fazalari miqdori 68,4% va anortit 31,6% ni tashkil qiladi.

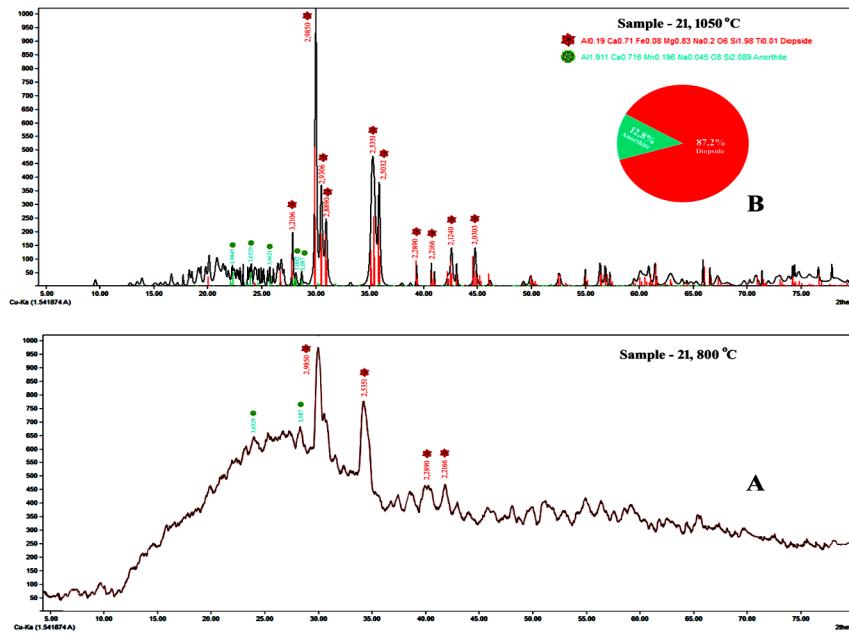
800 °C va 1050 °C haroratlarda 21-tarkibli shisha namunalariga issiqlik bilan ishlov berish jarayonida 2-tarkibli shisha namunaga nisbatan diopsid bilan bog‘liq bo‘lgan fazalar intensivligining oshishi kuzatildi. 3-rasmida olingan difraktogrammalardan aniq ko‘rinib turibdiki oldinroq ya’ni 800 °C haroratda diopsid ($d = 2, 2166; 2,2890; 2,5351; 2,9850 \text{ \AA}$) va anortit ($d = 3,087; 3,6529 \text{ \AA}$) hosil bo‘ladi.

Bazalt-kvarts-soda tizimida olingan shishakristall namunalarning fizik-kimyoviy xossalari quyidagi 4-jadvalda keltirilgan.

Shishakristall koshinlarning kristallanish darajasi ularning yuqori zichligi, shuningdek, past TCHKK qiymati bilan belgilanadi. Tahlil natijalari



2-rasm. 2-tarkibli shisha namunalarining A – 800 °C, B – 900 °C, V – 1050 °C haroratlarda kristallangan difraktogrammasi.



3-rasm. 21-tarkibli namunalarining A - 800 °C, B - 1050 °C haroratlardagi difraktogrammasi.

shuni ko'rsatdiki, Osmonsoy bazalti, Samarqand kvars qumi va Qo'ng'iroq sodasi asosida sintez qilingan shishani kristallash yo'li bilan olingan shishakristall namunalarining zichligi 2940 – 3100 kg/m³, TCHKK qiymati 53 – 64 K⁻¹ oralig'ida. Ushbu ko'rsatkichlar namunalarning yetarlicha yuqori kristallanish darajasini ko'rsatadi. Shisha texnologiyasidan foydalangan holda olingan shisha kristall namunalarining egilish kuchi 108-120 MPa, bosim kuchi esa 700-810 MPa oralig'ida. 3-jadvaldagagi ma'lumotlarda, kristallangan shisha

svuning yutuvchanligi - 0,009% dan kam, yuqori sovuqqa bardoshliligi - kamida 112 tsikl va yuqori kislotaga bardoshliligi (HCl kons.: 97,52-99,94%; H₂SO₄ kons.: 96,80-99,89%) keltirilgan.

Xulosa

Mahalliy xomashyolar asosida bazalt-kvars-soda tizimida bo'yovchi qo'shimchalardan foydalanmagan holda jigarrang, qora, firuza, yashil, och havorang va sariq rangli shisha namunalari sintez qilingan.

4-jadval

Shishakristall namunalarining fizik-kimyoviy xossalari

№	Xossalalar	Tarkiblar						GOST 13996-2019
		2	3	9	20	21	22	
1	Zichlik, kg/m ³	3089	3038	3010	2987	3100	2940	–
2	TCHKK•10 ⁷ , K ⁻¹	55	57	58	60	53	64	–
3	Mexanik egilishga mustahkamlik darajasi, MPa	119	116	113	111	120	108	15-35
4	Mexanik siqilishga mustahkamlik darajasi, MPa	794	762	745	720	810	700	–
5	Suv yutuvchanlik, %	0,003	0,003	0,005	0,007	0,001	0,009	0,5
6	Sovuqqa bardoshlilik, tsikl	137	134	129	116	145	112	100
7	Kislotala rga	konst. HCl	99,49	98,68	98,45	98,07	99,94	97,52
	bardoshlilik, %	konst. H ₂ SO ₄	98,51	98,34	98,16	97,23	99,89	96,80

Sintez qilingan rangli shishalarning zichligi 2481-2766 kg/m³, sovuqqa bardoshliligi 34-40 tsikl, mexanik egilishga mustahkamlik darajasi 33-55 MPani tashkill etib, GOST 54169-2010 talablarini qanoatlantirdi.

Bazalt-kvarts-soda tizimida sintez qilingan shishalarning kristallanish jarayoni ikki bosqichdan iborat bo‘lib, birinchi bosqich 800 °C va ikkinchi bosqich 1050 °C haroratni tashkil etadi.

Shishalarning ikki bosqichda kristallanishi

natijasida diopsid va anortit fazalarining hosil bo‘lishi aniqlandi.

Olingan shishakristall namunalarning zichligi 2940-3100 kg/m³, termik chiziqli kengayish koefisiyenti 53-64 K⁻¹, mexanik egilishga mustahkamlik darajasi 108-120 MPa, mexanik egilishga mustahkamlik darajasi 700-810 MPa, suv yutuvchanligi 0,001-0,009%, sovuqqa bardoshliligi 112-145 sikl bo‘lib, ushbu ko‘rsatichilar GOST 13996-2019 talablariga mos keldi.

REFERENCES

1. Shelby J.E. *Introduction to Glass Science and Technology*, 2nd ed., The Royal Society of Chemistry, Cambridge, 2005, 312 p.
2. Aripova M. Kh., Ruzmatov E.I. Sintez dekorativnogo stekla na osnove bazal’ta Omsonsayskogo mestorozhdeniya [Synthesis of decorative glass based on basalt from the Omsonsay deposit]. *Chemistry and Chemical Engineering*, 2022, 4, 3–8.
3. Kosic T., Krstic-Furundzic A.D. Architectural aspect of structural glass roof design. *Mid-term Conference “Structural Glass”*. Florida, 2013, 46–48.
4. Zhdanova A.S., Mazhirina A.D. Vidy stekla v arkhitekture [Types of glass in architecture]. *Shag v nauku*, 2019, 4, 21–24.
5. Magay A.A. Innovatsionnyye tekhnologii v osteklenii fasadov vysotnykh zdaniy [Innovative technologies in glazing the facades of high-rise buildings]. *Energosovet*, 2012, 4(23), 48–52.
6. Dayneko V.V., Kalikhman A.D.. Khudozhestvennyye izdeliya iz stekla v tekhnologii spekaniya dlya dekorativnogo oformleniya inter’yerov [Art glass products using sintering technology for interior decoration]. *Izvestiya vuzov: Investitsii. Stroitel’stvo. Nedyzhimost’*, 2013, 2(5), 144–151.
7. Pariafsai F. A review of design considerations in glass buildings. *Frontiers of Architectural Research*, 2016, 5(2), 171–193.
8. Sidikova T.D. Tekhnologiya polucheniya stekla i steklokristallicheskikh materialov s primeneniyem otkhodov proizvodstva [Technology for producing glass and glass-crystalline materials using production waste]. *Universum: Khimiya i biologiya*. 2018. 46(4), 13–15.
9. Sarkisov P.D., Orlova L.A., Popovich N.V. Current state of the matter in the field of technology and production of glass based aluminosilicate systems. Glass Formation, Crystallization and Formation in Preparation of Strontium Anorthite and Celcine Sital Stalks. *All Materials*, 2011, 8, 1–19.
10. Pavlushkin N.M. *Osnovy tehnologii sitallov* [The fundamentals of sital technology]. Moscow, Stroyizdat Publ., 1979. 340.
11. Manankov A.V., Strakhov B.S. Development of new composite building structures based on rock glass ceramics in creating industrial and transport infrastructure of oil and gas in the Arctic. *Vestnik TGASU*, 2014, 2, 167–176.
12. Kim A.YU., Kharitonov S.P., Amoyan M. Sozdaniye tekhnologicheskoy linii po proizvodstvu sitallovyykh izdeliy v usloviyah malogo predpriyatiya pri VUZe [Creation of a technological line for the production of glass-ceramic products in a small enterprise at a university]. *Innovatsii v nauke*. 2015, 41, 44–51.
13. Manankov A.V., Gasanova E.R. Sitally iz mestnogo syr’ya dlya proizvodstvennykh innovatsionnykh infrastruktur s vysokoy tekhniko-ekonomicheskoy effektivnost’yu v ekstremal’nykh usloviyah severa [Sitalls from local raw materials for production innovative infrastructures with high technical and economic efficiency in the extreme conditions of the Far North]. *Izvestiya Tomskogo politekhn. un-ta. Inzhiniring georesursov*. 2018, 329(11), 87–96.
14. Manankov A.V., Vladimirov V.M., Strakhov B.S. Rock glass_ceramics high strength design for working in special conditions in the Arctic. *Bulletin of TSU*, 2014, 386, 223–233.
15. Zhumaniyozov KH.P., Ismatov A.A., Sharipov D., Khodzhayev N.T. Poluchenije steklokristallicheskikh materialov na osnove diabazovykh gornykh porod Uzbekistana [Preparation of glass-crystalline materials based on diabase rocks of Uzbekistan]. *Khimicheskaya promyshlennost’*, 2011, 88(3), 120–125.
16. Aripova M.KH., Babakhanova Z.A., Zhumaniyozov KH.P. Steklokristallicheskiye plitki dlya polov na osnove mestnogo syr’ya i otkhodov promyshlennosti [Glass-crystalline floor tiles based on local raw materials and industrial waste]. *Universum: Tekhnicheskiye nauki*, 2020. 75(6), 76–80.
17. Willhauk E., Harikantha R. Glass ceramics for household appliances. In: Bach H., Krause D. (eds) *Low thermal expansion glass-ceramics*. 2nd edn. Springer Verlag, Heidelberg, 2005, 51–58. DOI: 10.1007/3-540-28245-9_3
18. Shishkin R.A., Yuferov Y.V., Polyvoda D.O. Fabrication of Basalt Matrix Composite Material by Pressureless Aluminum Melt Infiltration in Air Atmosphere. *Ceramics*, 2022, 780–788. DOI: 10.3390/ceramics5040056
19. Yilmaz S., Bayrak G., Sen S., Sen U. Structural characterization of basalt-based glass-ceramic coatings. *Materials & Design*, 2006, 27, 1092–1096.
20. Bayrak, Gunhan, Ediz Ercenk, Uğur Şen, Senol Yilmaz. Bond Strength of Basalt Based Glass-Ceramic Coatings. *Acta Physica Polonica A*, 2014, 125, 620–622.
21. Ruzmatov E. I., Aripova M. Kh. Razrabotka sostavov tsvetnykh stekol na osnove bazal’ta mestorozhdeniya Omsonsay [Development of compositions of colored glasses based on basalt from the Omsonsay deposit]. *Steklo i keramika*, 2024, 97(5), 11–16. DOI: 10.14489/glc.2024.05.pp.011-016
22. DOUGLAS P.E. et al. Optimization of glass-ceramic crystallization based on DTA exotherm analysis. *Dental materials: official publication of the Academy of Dental Materials*, 1994, 10(3), 167–171.